

# MATERIAL SAFETY DATA SHEET SDS/MSDS

# Section 1 - CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

### **PRODUCT NAME**

STANDARD SOLUTION "CERTISOL" AAS STRONTIUM 1000 MG/L IN NITRIC ACID NIST TRACEABLE

#### OTHER NAMES

"Strontium Standard Solution Traceable to SRM from NIST Sr(NO3)2 in HNO3 0.5 mol/L 1000 mg/L Sr CertiPUR."

### PROPER SHIPPING NAME

CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S.(contains nitric acid)

# **PRODUCT USE**

Reagent for analysis.

### **SUPPLIER**

BIO-CHEM Chemical 5455, Nicholson Road Science Market, Ambala Cantt. 133001 Haryana (India) +91-82952 41953 info@biofinechemical.com - www.biofinechemical.com

# **Section 2 - HAZARDS IDENTIFICATION**

### **GHS Classification**

Acute Toxicity Category 3
Eye Irritation Category 2A
Metal Corrosion Category 1
Skin Corrosion/Irritation Category 2





# **EMERGENCY OVERVIEW**

# **HAZARD**

**DANGER** 

Determined by using GHS criteria

H290 May be corrosive to metals.
H315 Causes skin irritation.
H319 Causes serious eye irritation.
H331 Toxic if inhaled.

# PRECAUTIONARY STATEMENTS

Prevention

Code Phrase

P234 Keep only in original container.

P261 Avoid breathing dust/ fume/ gas/ mist/ vapours/ spray.

P264 Wash ... thoroughly after handling.

P271 Use only outdoors or in a well- ventilated area.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

Response Code Phrase

P302+P352 IF ON SKIN: Wash with plenty of soap and water.

P304+P340 IF INHALED: Remove victim to fresh air and keep at rest in a position

comfortable for breathing.

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove

contact lenses, if present and easy to do. Continue rinsing.

P311 Call a POISON CENTER or doctor/physician.
P332+P313 If skin irritation occurs: Get medical advice/attention.
P337+P313 If eye irritation persists: Get medical advice/attention.
P362 Take off contaminated clothing and wash before re- use.

Absorb spillage to prevent material damage.

P390 Storage

Code Phrase

P403+P233 Store in a well- ventilated place. Keep container tightly closed.

P405 Store locked up.

P406 Store in corrosive resistant container or with a resistant inner liner.

Disposal Code Phrase

P501 Dispose of contents/container to ...

# Section 3 - COMPOSITION / INFORMATION ON INGREDIENTS

NAME CAS RN % 7697-37-2 1-5

# **Section 4 - FIRST AID MEASURES**

#### **SWALLOWED**

- For advice, contact a Poisons Information Centre or a doctor at once.
- Urgent hospital treatment is likely to be needed.
- If swallowed do NOT induce vomiting.
- If vomiting occurs, lean patient forward or place on left side (head-down position, if possible) to maintain open airway and prevent aspiration.

#### FYF

- If this product comes in contact with the eyes:
- Immediately hold eyelids apart and flush the eye continuously with running water.
- Ensure complete irrigation of the eye by keeping eyelids apart and away from eye and moving the eyelids by occasionally lifting the upper and lower lids.
- · Continue flushing until advised to stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.
- Transport to hospital or doctor without delay.

#### SKIN

- If skin or hair contact occurs:
- Immediately flush body and clothes with large amounts of water, using safety shower if available.
- Quickly remove all contaminated clothing, including footwear.
- Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre.
- Transport to hospital, or doctor.

# **INHALED**

- If fumes or combustion products are inhaled remove from contaminated area.
- Lay patient down. Keep warm and rested.
- Prostheses such as false teeth, which may block airway, should be removed, where possible, prior to initiating first aid procedures.
- Apply artificial respiration if not breathing, preferably with a demand valve resuscitator, bag-valve mask device, or pocket mask as trained. Perform CPR if necessary.

- Inhalation of vapours or aerosols (mists, fumes) may cause lung oedema.
- Corrosive substances may cause lung damage (e.g. lung oedema, fluid in the lungs).
- As this reaction may be delayed up to 24 hours after exposure, affected individuals need complete rest (preferably in semi-recumbent posture) and must be kept under medical observation even if no symptoms are (yet) manifested.
- Before any such manifestation, the administration of a spray containing a dexamethasone derivative or beclomethasone derivative may be considered.

### **NOTES TO PHYSICIAN**

- For acute or short term repeated exposures to strong acids:
- · Airway problems may arise from laryngeal edema and inhalation exposure. Treat with 100% oxygen initially.
- · Respiratory distress may require cricothyroidotomy if endotracheal intubation is contraindicated by excessive swelling
- Intravenous lines should be established immediately in all cases where there is evidence of circulatory compromise.
- Strong acids produce a coagulation necrosis characterised by formation of a coagulum (eschar) as a result of the dessicating action of the acid on proteins in specific tissues.

# **Section 5 - FIRE FIGHTING MEASURES**

### **EXTINGUISHING MEDIA**

- · Water spray or fog.
- Foam.
- Dry chemical powder.
- · BCF (where regulations permit).

### **FIRE FIGHTING**

- · Alert Fire Brigade and tell them location and nature of hazard.
- Wear full body protective clothing with breathing apparatus.
- Prevent, by any means available, spillage from entering drains or water course.
- Use fire fighting procedures suitable for surrounding area.

# FIRE/EXPLOSION HAZARD

- · Non combustible.
- · Not considered to be a significant fire risk.
- · Acids may react with metals to produce hydrogen, a highly flammable and explosive gas.
- Heating may cause expansion or decomposition leading to violent rupture of containers.

### FIRE INCOMPATIBILITY

■ None known.

# **Section 6 - ACCIDENTAL RELEASE MEASURES**

### MINOR SPILLS

- Drains for storage or use areas should have retention basins for pH adjustments and dilution of spills before discharge or disposal of material.
- · Check regularly for spills and leaks.
- · Clean up all spills immediately.
- · Avoid breathing vapours and contact with skin and eyes.
- Control personal contact with the substance, by using protective equipment.
- Contain and absorb spill with sand, earth, inert material or vermiculite.

### **MAJOR SPILLS**

- Clear area of personnel and move upwind.
- Alert Fire Brigade and tell them location and nature of hazard.
- Wear full body protective clothing with breathing apparatus.
- Prevent, by any means available, spillage from entering drains or water course.

Personal Protective Equipment advice is contained in Section 8 of the MSDS.

# Section 7 - HANDLING AND STORAGE

### PROCEDURE FOR HANDLING

- DO NOT allow clothing wet with material to stay in contact with skin.
- · Avoid all personal contact, including inhalation.
- · Wear protective clothing when risk of exposure occurs.
- · Use in a well-ventilated area.
- · Avoid contact with moisture.

#### **SUITABLE CONTAINER**

- DO NOT use aluminium or galvanised containers.
- · Check regularly for spills and leaks.
- · Lined metal can, lined metal pail/ can.
- · Plastic pail.
- Polyliner drum.
- · Packing as recommended by manufacturer.

For low viscosity materials

- Drums and jerricans must be of the non-removable head type.
- Where a can is to be used as an inner package, the can must have a screwed enclosure. <</>>

#### STORAGE INCOMPATIBILITY

- Inorganic acids are generally soluble in water with the release of hydrogen ions. The resulting solutions have pH's of less than 7.0.
- Inorganic acids neutralise chemical bases (for example: amines and inorganic hydroxides) to form salts neutralisation can generate dangerously large amounts of heat in small spaces.
- The dissolution of inorganic acids in water or the dilution of their concentrated solutions with additional water may generate significant heat.
- The addition of water to inorganic acids often generates sufficient heat in the small region of mixing to cause some of the water to boil explosively. The resulting "bumping" can spatter the acid.
- Reacts with mild steel, galvanised steel / zinc producing hydrogen gas which may form an explosive mixture with air.
- Avoid strong bases.

# STORAGE REQUIREMENTS

Keep containers securely sealed.
 Store between 15 and 25 deg. C.

### Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

# EXPOSURE CONTROLS Source Material TWA ppm TWA mg/m³ STEL ppm STEL mg/m³ India Permissible nitric acid (Nitric acid) Levels of Certain acid) Chemical Substances in

#### **MATERIAL DATA**

Work Environment

MERCK STRONTIUM STANDARD SOLUTION TRACEABLE TO SRM FROM: NITRIC ACID:

■ For nitric acid:

Odour Threshold Value: 0.27 ppm (detection)

NOTE: Detector tubes for nitric acid, measuring in excess of 5 ppm, are commercially available.

The TLV-TWA is protective against corrosion of the skin, tissue and other membranes, against irritation to the eyes and mucous membranes, and against acute pulmonary oedema or chronic obstructive lung disease.

### PERSONAL PROTECTION

# STANDARD SOLUTION "CERTISOL" AAS STRONTIUM 1000 MG/L IN NITRIC ACID NIST TRACEABLE.

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### Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION









#### RESPIRATOR

•Type AE-P Filter of sufficient capacity. (AS/NZS 1716 & 1715, EN 143:2000 & 149:2001, ANSI Z88 or national equivalent)

### **EYE**

- · Chemical goggles.
- Full face shield may be required for supplementary but never for primary protection of eyes
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. A written policy document, describing the wearing of lens or restrictions on use, should be created for each workplace or task. This should include a review of lens absorption and adsorption for the class of chemicals in use and an account of injury experience. Medical and first-aid personnel should be trained in their removal and suitable equipment should be readily available. In the event of chemical exposure, begin eye irrigation immediately and remove contact lens as soon as practicable. Lens should be removed at the first signs of eye redness or irritation lens should be removed in a clean environment only after workers have washed hands thoroughly. [CDC NIOSH Current Intelligence Bulletin 59], [AS/NZS 1336 or national equivalent].

### HANDS/FEET

- Wear chemical protective gloves, e.g. PVC.
- Wear safety footwear or safety gumboots, e.g. Rubber.
- · When handling corrosive liquids, wear trousers or overalls outside of boots, to avoid spills entering boots.

The selection of the suitable gloves does not only depend on the material, but also on further marks of quality which vary from manufacturer to manufacturer. Where the chemical is a preparation of several substances, the resistance of the glove material can not be calculated in advance and has therefore to be checked prior to the application.

The exact break through time for substances has to be obtained from the manufacturer of the protective gloves and has to be observed when making a final choice.

Suitability and durability of glove type is dependent on usage. Important factors in the selection of gloves include:

# **OTHER**

- Overalls.
- PVC Apron.
- PVC protective suit may be required if exposure severe.
- Eyewash unit.

# **ENGINEERING CONTROLS**

■ Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection.

The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

Enclosure and/or isolation of emission source which keeps a selected hazard "physically" away from the worker and ventilation that strategically "adds" and "removes" air in the work environment.

# Section 9 - PHYSICAL AND CHEMICAL PROPERTIES

# **APPEARANCE**

Colourless liquid.

### **PHYSICAL PROPERTIES**

Liquid. Mixes with water. Corrosive.

Acid.

State
Melting Range (°C)
Boiling Range (°C)
Flash Point (°C)
Decomposition Temp (°C)
Autoignition Temp (°C)

Liquid Not Available Not Available Not Applicable Not Available Not Available

Molecular Weight Viscosity Solubility in water (g/L) pH (1% solution) pH (as supplied) Vapour Pressure (kPa) Not Applicable Not Available Miscible Not Available 0.5 approx @ 20C Not Available Upper Explosive Limit (%) Lower Explosive Limit (%) Not Applicable Not Applicable Specific Gravity (water=1) Relative Vapour Density

(air=1)

**Evaporation Rate** 

Volatile Component (%vol) Not Available

Not Available

Not Available

1.02 @ 20C

# Section 10 - CHEMICAL STABILITY AND REACTIVITY INFORMATION

# **CONDITIONS CONTRIBUTING TO INSTABILITY**

· Contact with alkaline material liberates heat.

For incompatible materials - refer to Section 7 - Handling and Storage.

# Section 11 - TOXICOLOGICAL INFORMATION

## Health hazard summary table:

Acute toxicity Acute Tox. (inhal) 3 Skin Irrit. 2 Skin corrosion/irritation Serious eye damage/irritation Eye Irrit. 2A Respiratory or skin sensitization Not applicable Germ cell mutagenicity Not applicable Carcinogenicity Not applicable Reproductive toxicity Not applicable STOT- single exposure Not applicable STOT- repeated exposure Not applicable Aspiration hazard Not applicable

### POTENTIAL HEALTH EFFECTS

# **ACUTE HEALTH EFFECTS**

#### **SWALLOWED**

■ Ingestion of acidic corrosives may produce burns around and in the mouth, the throat and oesophagus. Immediate pain and difficulties in swallowing and speaking may also be evident.

#### EYE

- Direct eye contact with acid corrosives may produce pain, tears, sensitivity to light and burns.
- Mild burns of the epithelia generally recover rapidly and completely.
- This material can cause eye irritation and damage in some persons.

# SKIN

- Open cuts, abraded or irritated skin should not be exposed to this material.
- Skin contact with acidic corrosives may result in pain and burns; these may be deep with distinct edges and may heal slowly with the formation of scar tissue.
- Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.
- This material can cause inflammation of the skin on contact in some persons.

# **INHALED**

■ Corrosive acids can cause irritation of the respiratory tract, with coughing, choking and mucous membrane damage. There may be dizziness, headache, nausea and weakness.

# **CHRONIC HEALTH EFFECTS**

■ Substance accumulation, in the human body, may occur and may cause some concern following repeated or long-term occupational exposure.

Repeated or prolonged exposure to acids may result in the erosion of teeth, swelling and/or ulceration of mouth lining. Irritation of airways to lung, with cough, and inflammation of lung tissue often occurs. Chronic exposure may inflame the skin or conjunctiva.

# TOXICITY AND IRRITATION

■ Not available. Refer to individual constituents.

SKIN

nitric acid GESAMP/EHS Composite List - GESAMP Hazard

**Profiles** 

D1: skin

irritation/corrosion

3C

# Section 12 - ECOLOGICAL INFORMATION

This material and its container must be disposed of as hazardous waste.

**Ecotoxicity** 

Ingredient Persistence: Persistence: Air Mobility Bioaccumulation

Water/Soil

nitric acid No Data No Data LOW No Data Available

Available Available

# **Section 13 - DISPOSAL CONSIDERATIONS**

· Recycle wherever possible.

- · Consult manufacturer for recycling options or consult local or regional waste management authority for disposal if no suitable treatment or disposal facility can be identified.
- Treat and neutralise at an approved treatment plant. Treatment should involve: Neutralisation with soda-ash or soda-lime followed by: burial in a land-fill specifically licenced to accept chemical and / or pharmaceutical wastes or Incineration in a licenced apparatus (after admixture with suitable combustible material).
- Decontaminate empty containers with 5% aqueous sodium hydroxide or soda ash, followed by water. Observe all label safeguards until containers are cleaned and destroyed.

# Section 14 - TRANSPORTATION INFORMATION



Labels Required: CORROSIVE

## HAZCHEM:

Land Transport UNDG:

Class or division: Subsidiary risk: None UN No.: 3264 UN packing group:

Shipping Name: CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (contains

nitric acid)

Air Transport IATA:

ICAO/IATA Class: ICAO/IATA Subrisk: None UN/ID Number: 3264 Packing Group:

Special provisions: A3A803

Shipping name: CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (contains nitric acid)

**Maritime Transport IMDG:** 

IMDG Class: 8 IMDG Subrisk: None 3264 UN Number: Packing Group: Ш EMS Number: F- A, S- B Special provisions: 223 274

Limited Quantities: 5 L

Shipping name: CORROSIVE LIQUID, ACIDIC, INORGANIC, N.O.S. (contains nitric acid)

# **Section 15 - REGULATORY INFORMATION**

# **REGULATIONS**

### Regulations for ingredients

# nitric acid (CAS: 7697-37-2) is found on the following regulatory lists;

"Fisher Transport Information", "GESAMP/EHS Composite List - GESAMP Hazard Profiles", "IMO IBC Code Chapter 17: Summary of minimum requirements", "IMO MARPOL 73/78 (Annex II) - List of Noxious Liquid Substances Carried in Bulk", "India Hazardous Wastes (Management, Handling and Transboundary Movement) Rules - Schedule 2: List of Wastes Constituents with Concentration Limits", "India Manufacture, Storage and Import of Hazardous Chemical Rules - Schedule 1: List of Hazardous and Toxic Chemicals", "India Permissible Levels of Certain Chemical Substances in Work Environment", "International Council of Chemical Associations (ICCA) - High Production Volume List", "OECD List of High Production Volume (HPV) Chemicals"

# **Section 16 - OTHER INFORMATION**

- Classification of the preparation and its individual components has drawn on official and authoritative sources using available literature references.
- The (M)SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings.